

Classification of Amino acids

①

① Classification based on the structure

2) " " on polarity

" " nutrition

3) " " metabolic rate

4) " "

② Classification based on the structure.

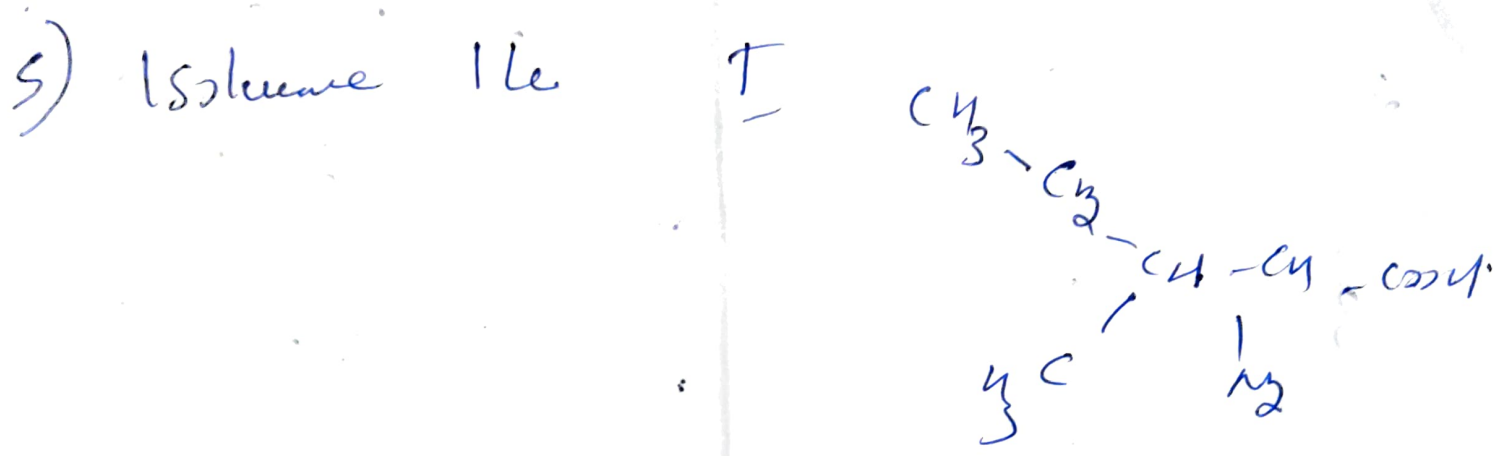
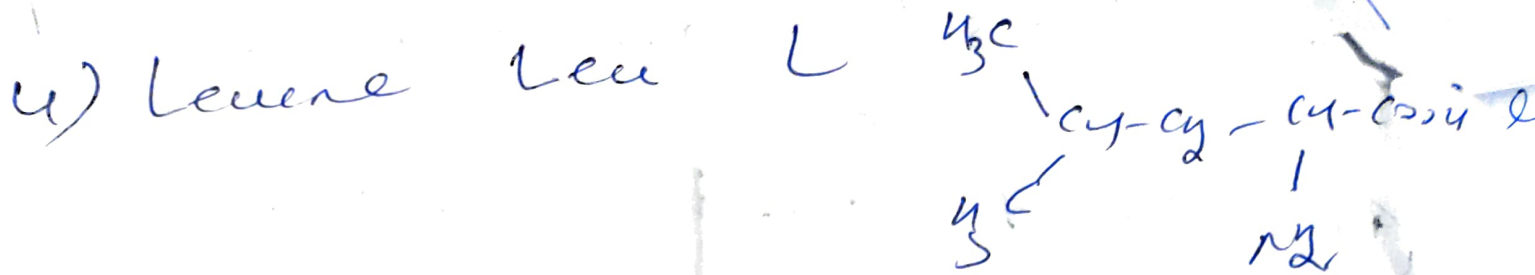
1. Amino acids with aliphatic side

Char's

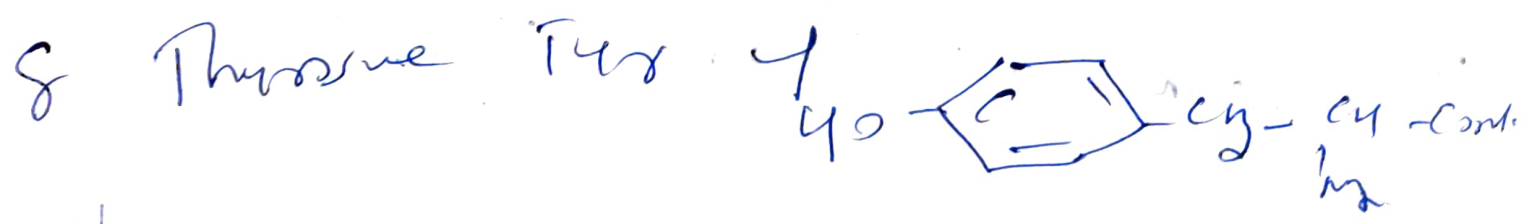
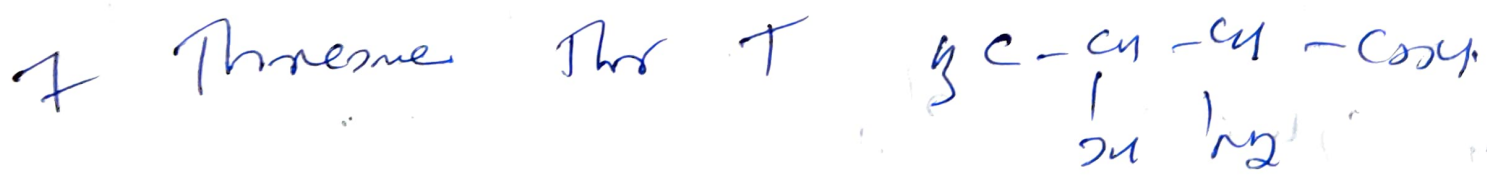
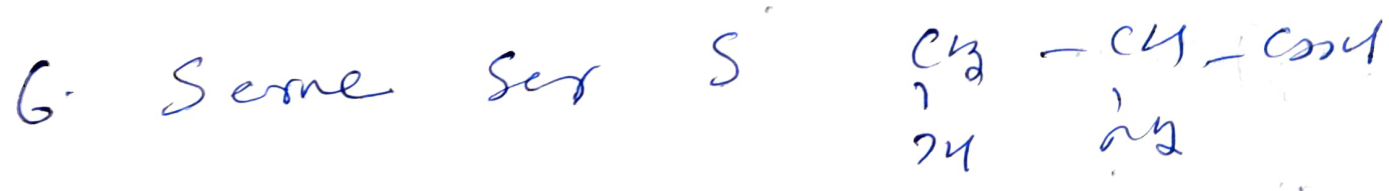
1. Glutamic acid Gly G $\text{H}-\underset{\text{NH}_2}{\text{CH}}-\text{COOH}$

2. Alanine Ala A $\text{CH}_3-\underset{\text{NH}_2}{\text{CH}}-\text{COOH}$

3. Valine Val V $\begin{array}{c} \text{H}_3\text{C} \\ \diagdown \\ \text{CH}-\underset{\text{NH}_2}{\text{CH}}-\text{COOH} \\ \diagup \\ \text{H}_3\text{C} \end{array}$

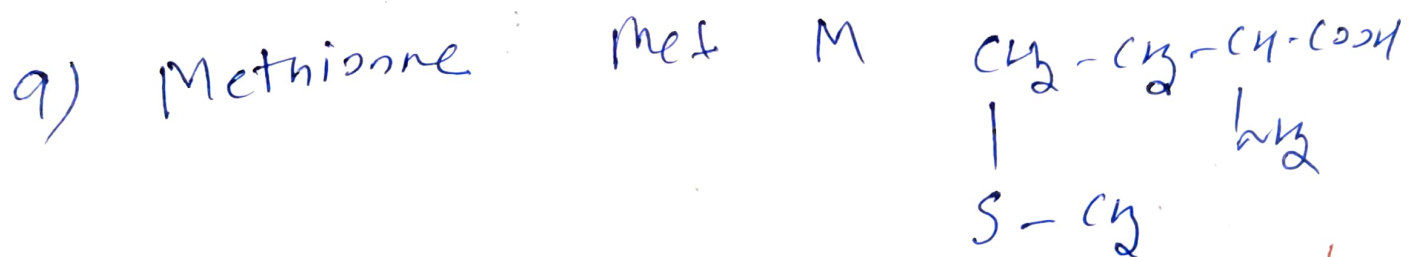
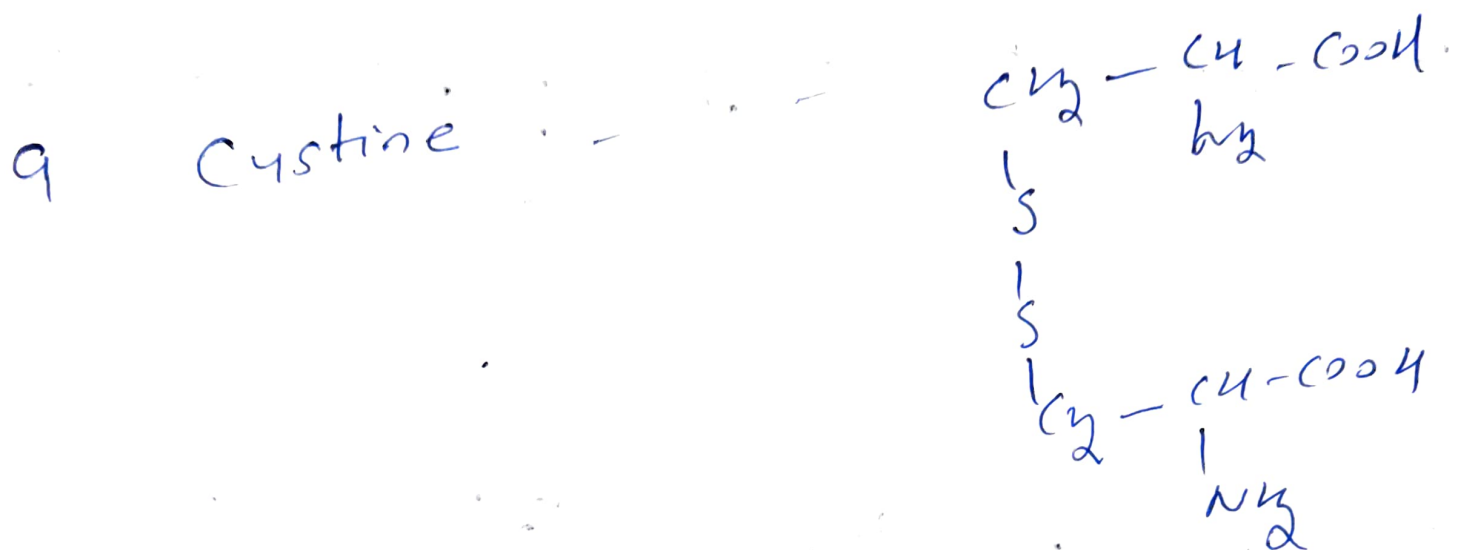
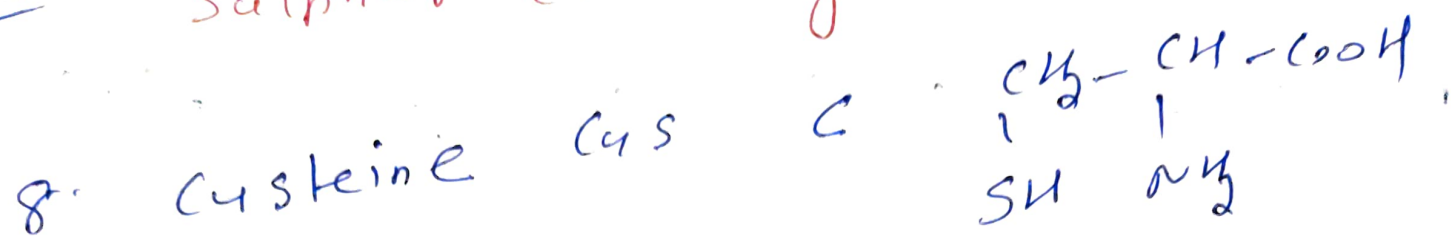


B Amino acids containing hydroxyl group

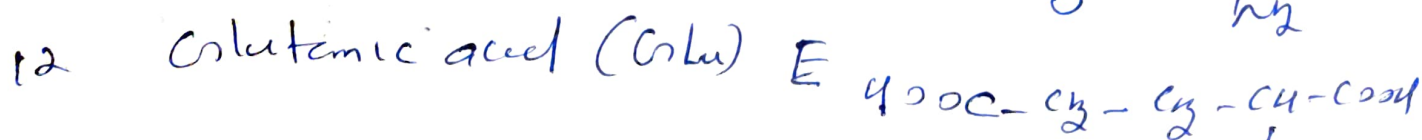
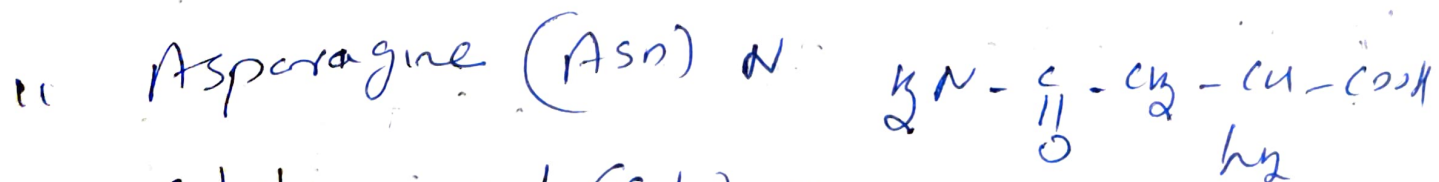


9)

Sulphur containing amino acids



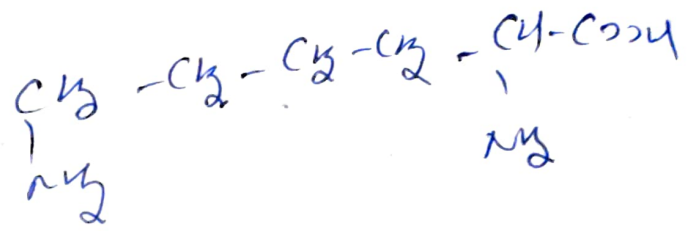
IV. Acidic Amino acids and their Amides



5. Basic Amino acids

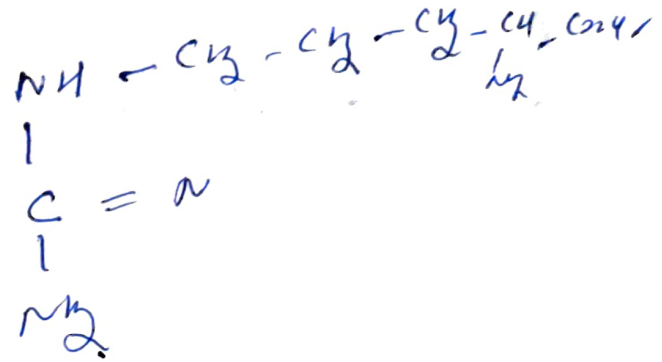
14. Lysine Lys

1c



15. Arginine

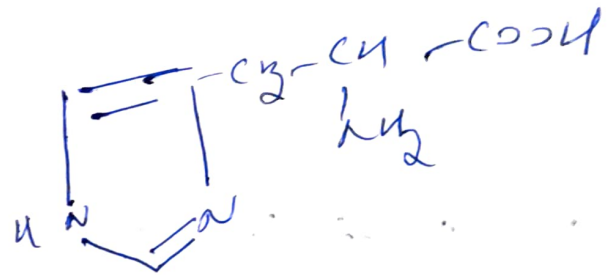
Arg. R



16. Histidine

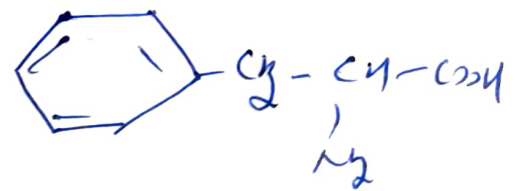
His

H



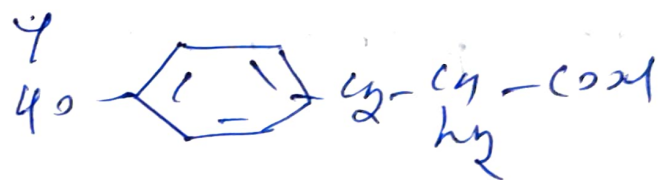
6. Aromatic Amino acids

17) Phenylalanine Phe F



18) Tyrosine

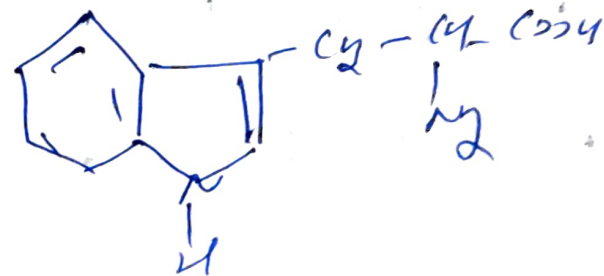
Tyr



19) Tryptophan

Trp

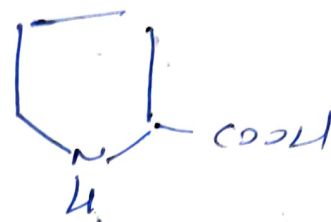
w



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III Amino acid

2. Proline



proline

Properties of Amino Acids

A. physical properties

B. Chemical properties

A. physical properties

1. Solubility — Soluble in water and

insoluble in organic solvents

2) Melting point — Melt at high temperature
at above 200°C

3. Taste - Amino acids may be sweet
eg. Cysteine, Valine

Tasteless - Leucine

Bitter - Arginine, Isoleucine

4. Optical properties

All the amino acids except + glycine
possess optical isomers due to the
presence of a symmetric carbon atom

5. Amino acids as ampholytes

→ Amino acids contain both acidic
and basic groups

→ They can donate a proton or
accept a proton, hence amino acids
are regarded as ampholytes.

u) Zwitterion or dipolar ion (u)

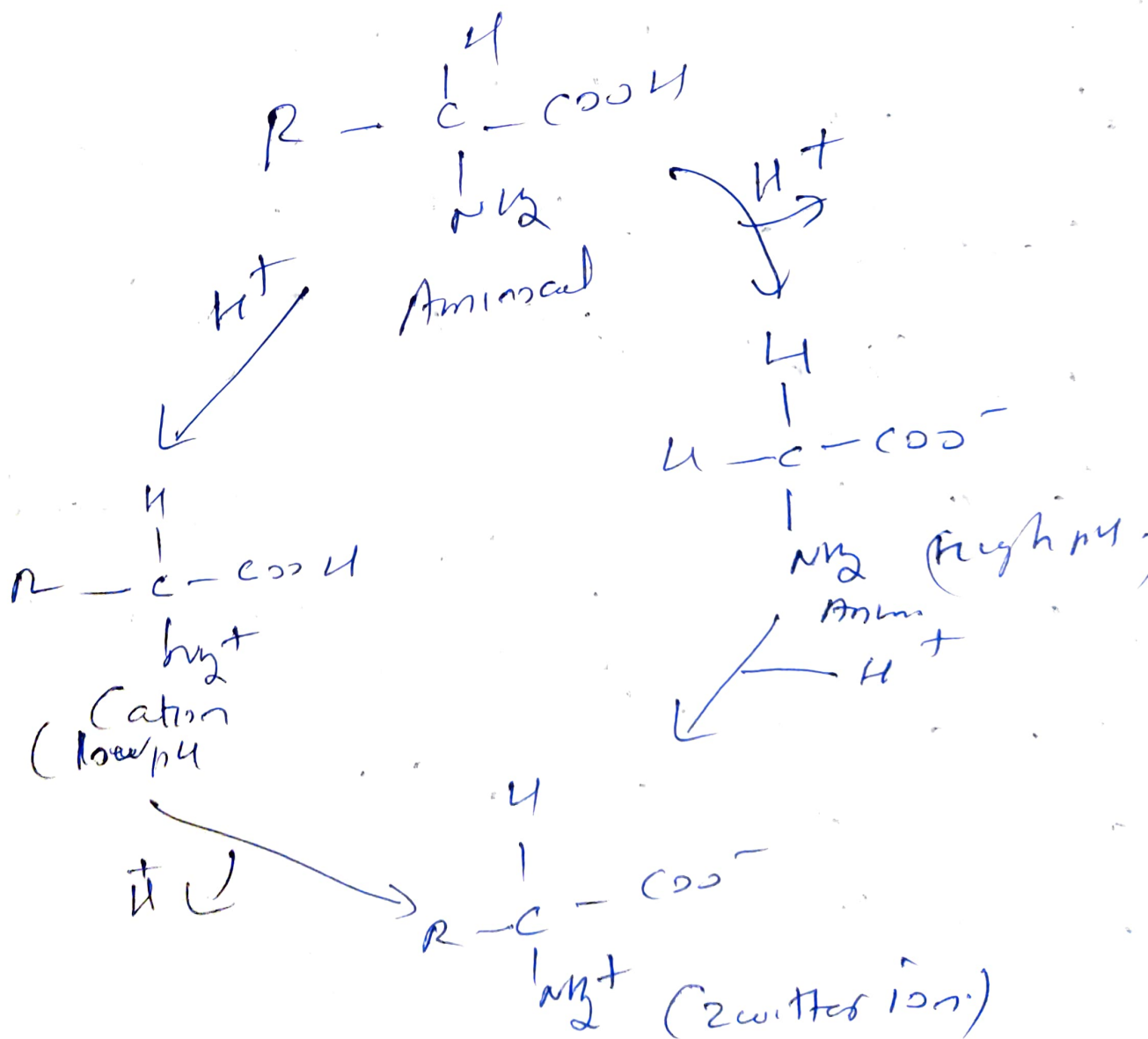
→ The amino acids rarely exist in a neutral form with free carboxylic

(COOH) and free amino ($-NH_2$) groups.

→ In strongly acidic pH (low pH) the amino acid is positively charged (cation) while in strongly alkaline pH (high pH), it is negatively charged (anion).

→ Each amino acid has a characteristic pH at which it carries both positive and negative charges and exists as zwitterion.

Isoelectric pH (pI) is defined
 as the pH at which a molecule
 exist as a zwitterion or dipolar
 ion and carries no net charge
 → thus, the molecule is electrologically
 neutral



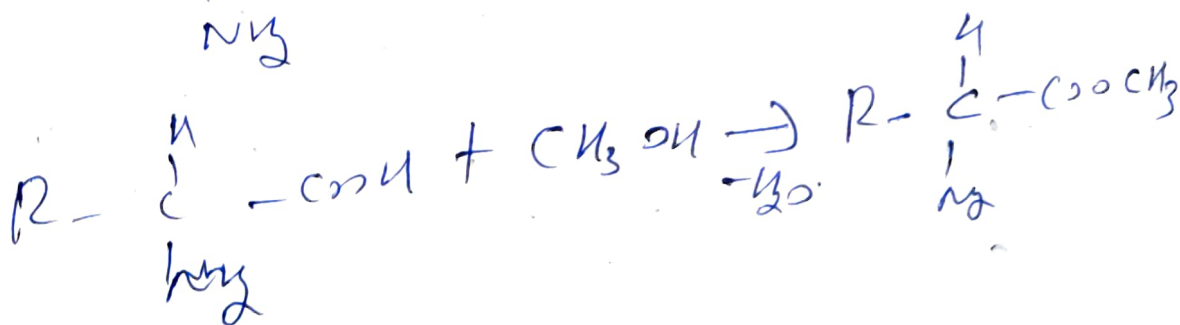
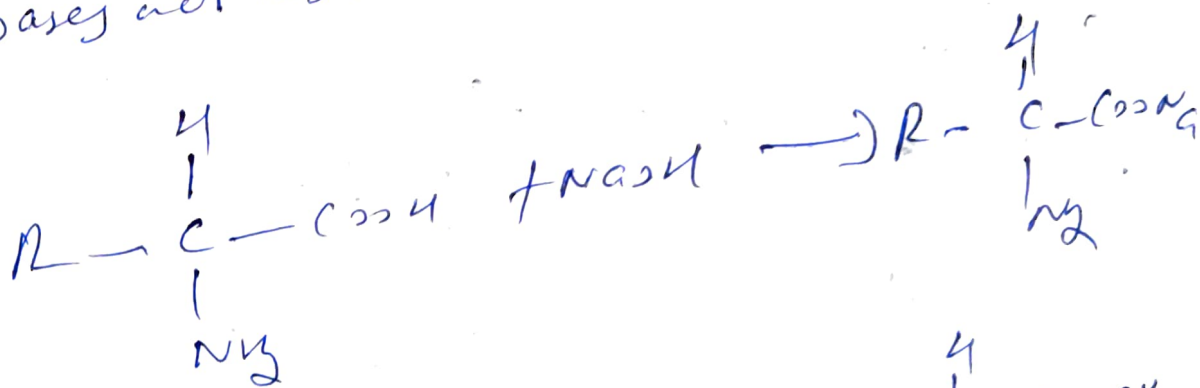
Chemical Properties

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→ The general reactions of amino acids are mostly due to the presence of two functional groups namely Carboxyl ($-COOH$) group and Amino ($-NH_2$) group.

Reactions due to $-COOH$ group.

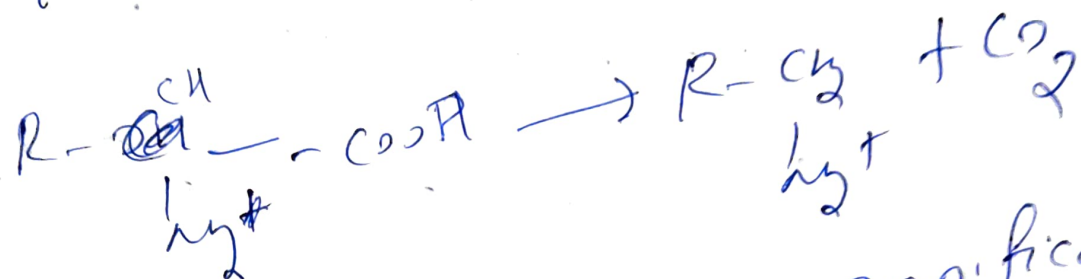
1. Amino acids form salts with bases and esters with alcohols.



2) Decarboxylation

Amino acids undergo decarboxylation.

to produce corresponding amines



→ This reaction assumes significance in the lung cells due to the formation of many biologically important amines

→ These include

histidine → histamine

Tyrosine → Tyramine

Glutamate → γ -aminobutyric acid

(6).

3) Reactions with ammonia

→ The carboxyl group of dicarboxylic amino acids reacts with NH_3 to form amide

Aspartic acid + NH_3 → Asparagine

Glutamic acid + NH_3 → Glutamine

II Reactions due to NH_2 group

→ The amino group behave as bases
Combine with acids (eg HCl) to form salts ($\text{NH}_3^+ \text{Cl}^-$)

→ Amino acids can be identified by specific colour reactions

-cont

3) Transamination

Transfer of an amino group from an amino acid to a keto acid to form a new amino acid is a very important reaction in amino acid metabolism

ii) Oxidative deamination

→ The amino acids undergo oxidative deamination to liberate free ammonia

Based on polarity

1) Non polar - Alanine, Leucine, Isoleucine, Valine, methionine, phenylalanine, Tryptophan, proline

2) polar - glycine, Serine, Threonine, Cysteine, glutamine, histidine

Based on nutrition

Essential amino acids

→ An essential amino acid that cannot be synthesized by the organism and must therefore come from the diet

They are

phenylalanine

Valine

Threonine

Tryptophan

Methionine

Isoleucine

Isolucine

Semi essential
Histidine
Lysine

Alanine

Arginine

Asparagine

Cysteine

Glutamic acid

Glucose

Proline

Selenocysteine

Serine

Threonine

Bond in metabolic fate

1) Catabolic amino acid

Alanine, aspartate, glycine,
methionine

2) Ketogenic

leucine and lysine

3) Catabolic and Ketogenic

Isoleucine, phenylalanine,

tyrosine, threonine